HIGHLY STEREOSELECTIVE APPROACH TO CHIRAL BUILDING BLOCK POSSESSING THREE CONTIGUOUS ASYMMETRIC CENTERS. PREPARATION OF FOUR POSSIBLE DIASTEREOMERS OF β , β '-DIMETHYL-BIS-HOMOALLYLIC ALCOHOL DERIVATIVE

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Summary: All four diastereomers of chiral alcohol 2 with three contiguous chiral centers were stereoselectively prepared via diastereoselective addition of nucleophiles (crotyl metal or H^-) to α -methyl- β , γ -unsaturated carbonyl compounds 1.

Stereo-control of acyclic molecules, both in the enantio- and diastereomerical senses, is a significant problem in organic synthesis. Recently, the potential of the organoaluminum-promoted pinacol-type rearrangement has been demonstrated for the enantiospecific synthesis of chiral building blocks.¹⁾ In connection with our research in the macrolide synthesis, our attention was focussed on the diastereoselective extention of the carbon framework of such chiral building blocks, which would lead to the advanced intermediates having multiple chiral centers. It is one of the common and essential problems in the macrolide synthesis where a wide variety of methods have been devised as exemplified in the "asymmetric aldol strategy" to overcome the so-called "Cram problem" of the chiral carbonyl compounds.²⁾ Our own clue to solve this problem is the exceptionally high diastereoselectivity in the nucleophilic addition to the α -methyl- β , γ -unsaturated carbonyl compounds 1.³⁾



In this communication, we wish to describe a highly stereo-controlled approach to such advanced intermediates 2 possessing three contiguous chiral centers. By the full use of the aforementioned high diastereoselectivity, there was realized a stereoselective access to all the four possible isomers of the chiral β , β '-dimethyl-bis-homoallylic alcohol 2.⁴

The aldehyde 6 was prepared from the chiral amide 3 (available from (S)ethyl lactate)¹⁾ via the reductive pinacol-type rearrangement^{1b)} followed by oxidation.⁵⁾ The overall yield was 64% and the geometry of the migrating C-3unit, 3-alkoxy-1-propenyl group, was retained throughout the process.



First, the stereoselectivity of the crotylation of 6 was examined: Crotylation of $\underline{6}$ with the Cr-mediated reagent (CrCl₃ - 1/2LiAlH₄, MeCH=CHCH₂Br / THF, 0°C)^{6b)} gave the isomer 2a in a highly stereoselective manner. On the other hand, the reaction under the Lewis acid-mediated conditions (Bu₃SnCH₂CH= CHMe, $BF_3 \cdot OEt_2 / CH_2Cl_2$, -78°C)^{6C)} resulted in the predominant formation of the isomer <u>2b</u>,⁷⁾ as shown in Table I.⁸⁾ These results reflect the following two features: (i) The C(4)-C(5) stereochemistry was controlled by the chirality of the aldehyde 6 (the Cram selectivity: 10/1, 12/1) as rationalized in terms of the Felkin-Anh's model^{3,9)} and (ii) the C(5)-C(6) counterpart was controlled by the diastereoselectivity inherent in the crotyl metal reagent employed.⁶⁾ The stereochemistry of the adduct 2a was confirmed by converting it to the Prelog-Djerassi lactonic acid and the related compound. 10,11,12)



Method A: MeCH=CHCH2Br, CrCl3-1/2 LiAlH4 /THF, 0°C. Method B: Bu₃SnCH₂CH=CHMe, BF₃·OEt₂ / CH₂Cl₂,

 $R = PhCH_0CH_0-$

2b

 $-78^{\circ}C$. The other isomers, <u>2c</u> and <u>2d</u>, were also made accessible in highly stereoselective manner by the inversion of the C(5)-hydroxyl of 2a or 2b, efficiently achieved by the following oxidation-reduction sequence (method C): Oxidation of 2a or 2b gave the corresponding ketones 7a or 7b, without any epimerization Subsequent reduction of these ketones with Super-Hydride or the conjugation. (LiBEt₃H / THF, -78°C) resulted in the exclusive formation of the alcohol 2c or



2d, respectively, and no trace of the parent alcohols, 2a and 2b, were detected in each of the cases.⁸⁾ Thus, the stereospecific conversion of the isomeric alcohols was realized, that is, from 2a to 2c, and from 2b to 2d.^{7,12)}

In addition to the apparent synthetic utility of the present reduction, it is also notable that the stereochemical outcome of the H attack totally depends on the chirality at C(4) irrespective of that at C(6). Thus, the stereochemical bias posed by the Me₃Si-substituted alkenyl group at C(4) greatly exceeds the non-substituted one at C(6), which could be well understood by our systematic investigation on the substituent dependence of the stereoselectivity in the reduction of the related model substrates as described earlier.^{3a)}

In Table I is summarized the whole feature of the present process, which depends on the highly diastereoselective attack of the nucleophilic species (crotyl metals or H⁻) onto the α -methyl- β , γ -unsaturated carbonyl compounds. By way of these direct and indirect methods, all of the four possible isomers of the β , β '-dimethyl-bis-homoallylic alcohols 2a - 2d were synthesized with high stereochemical purities in a flexible manner.¹³

Application of the present method to the stereoselective synthesis of chiral natural products is now extensively studied in our laboratory.¹¹⁾

Starting Material	Method ^{a)}				
		<u>2a</u>	<u>2b</u>	<u>2c</u>	<u>2d</u>
<u>6</u>	A ^{c)}	>99	<1		
<u>6</u>	B ^d)	4	96		
<u>2a</u>	C ^{f)}	not detected >99 <1		<1	
<u>2b</u> e)	c ^{f)}	not detected		4	96

Table I. Stereoselective Synthesis of Alcohol 2

a) See text. b) By HPLC, see ref 8). c) 78%; (2a+2b)/(2c+2d) = 10/1. d) 85%; (2a+2b)/(2c+2d) = 12/1. e) 2b containing 4% 2a was used, obtained by method B. f) Yield in two steps: >90%.

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References and Notes

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Isomer	[α] _D in CHCl₃	13 C NMR (δ , CDC1 ₃)
<u>2a</u>	$[\alpha]_{D}^{25}$ -2.5° (c 2.5).	0.76, 14.6, 16.9, 40.8, 41.3, 67.2, 69.5, 76.2, 94.2,
<u>2b</u>	$[\alpha]_D^{24}$ -2.6° (c 2.6).	0.73, 13.4, 16.1, 40.8, 41.3, 67.3, 69.5, 75.7, 94.2,
<u>2c</u>	$[\alpha]_D^{23}$ -20° (c 0.86).	114.6, 127.7, 127.9, 128.4, 137.9, 138.6, 141.4, 147.6 0.60, 11.1, 18.4, 39.4, 41.6, 67.3, 69.6, 78.3, 94.3,
<u>2d</u>	$[\alpha]_{D}^{23}$ -32° (c 0.94).	113.9, 127.7, 127.9, 128.4, 138.0, 138.1, 142.9, 148.2 0.57, 18.3, 18.4, 40.4, 42.6, 67.3, 69.6, 78.8, 94.3, 115.4, 127.7, 127.9, 128.4, 137.9, 138.4, 139.2, 148.0

7) The $[\alpha]_{p}$ values and ¹³C NMR spectra of isomers 2a - 2d are as follows;

- 8) On SiO₂ TLC (hexane/Et₂O = 7/3), a mixture of 2a 2d was easily separated into two spots, where the lower spot (Rf 0.29) consisted of $\frac{2a}{2}$ and $\frac{2b}{2}$, and the higher one (Rf 0.43) consisted of $\frac{2c}{2}$ and $\frac{2d}{2}$. Ratio of $\frac{2a}{2b}$, and that of $\frac{2c}{2d}$ were determined by the HPLC analysis of the separated spots stated above. HPLC conditions are as follows; 2a/2b: Develosil ODS-5 (4.6/250, Nomura Chem., Co.), MeOH/H₂O = 4/1. 2c / 2d: ZORBAX SIL (4.6/250, Du Pont Instruments), hexane / AcOEt = 20 / 1.
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